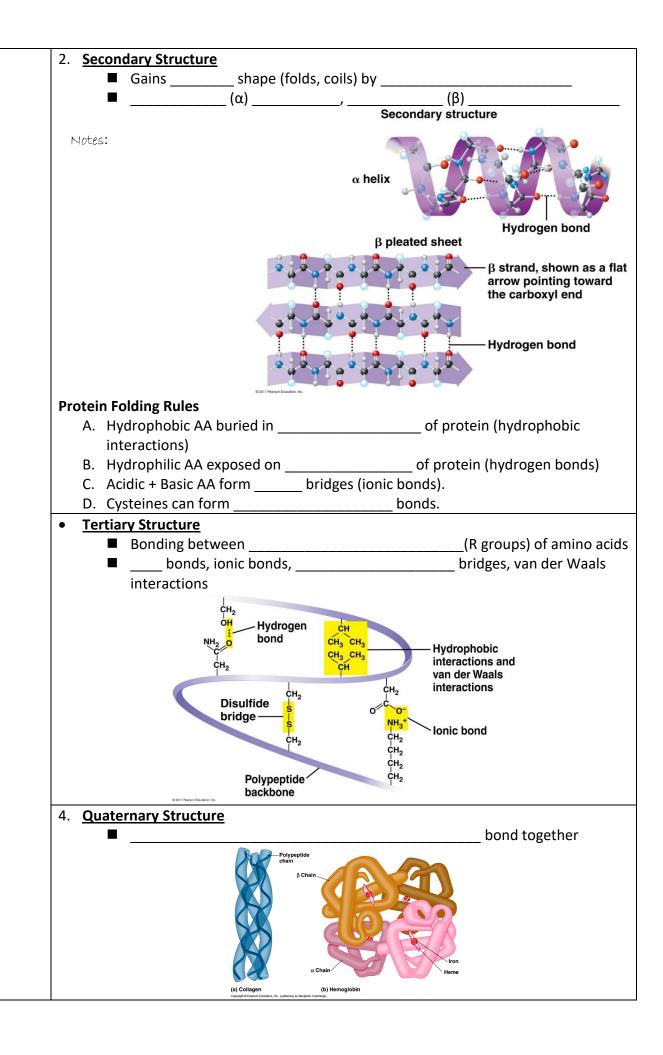
**Ch. 5: The Structure and Function of Large Biological Molecules** 

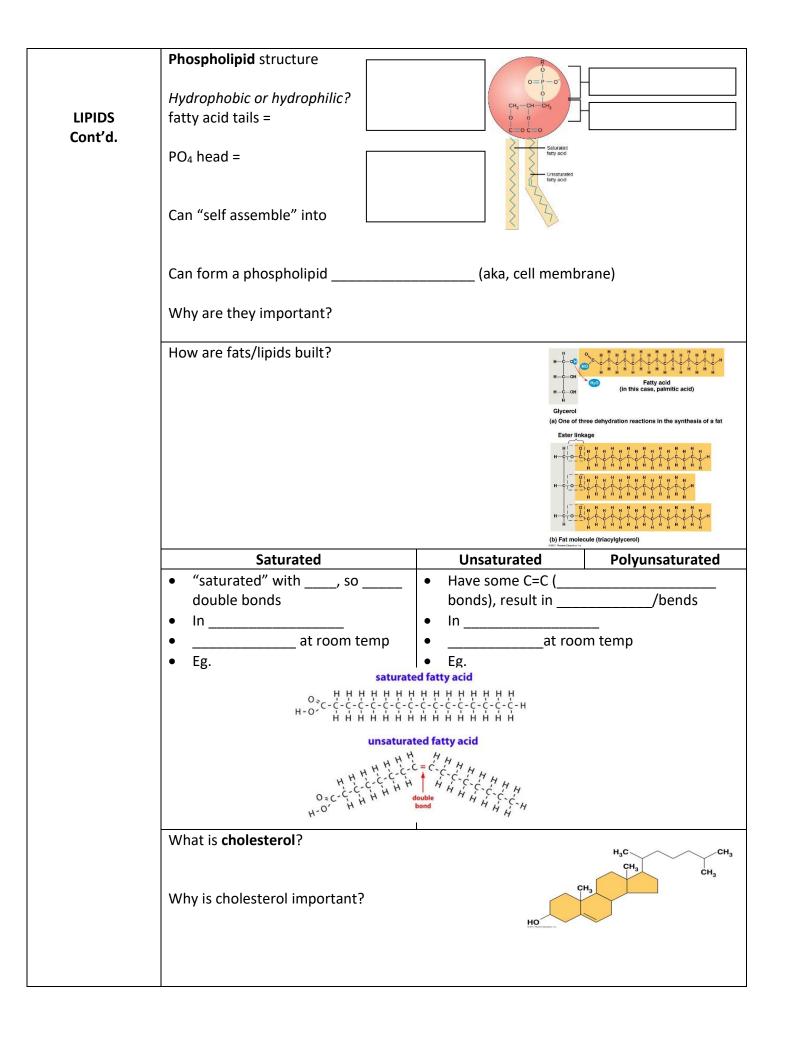
	Monomers		Polymers	Macromolecules
Vocabulary	<ul> <li>Used for b of polymers</li> <li>Connects with reaction =</li> </ul>		molecules of many identical or r blocks linked by	<ul> <li>molecules</li> <li>2 or more polymers bonded together</li> </ul>
Types of Reactions	Dehydration Synthes (Condensation Reaction  Make  A + B →	on)	• • AB →	Hydrolysis polymers _ →
	(a) Dehydration reaction: synthesizing a polymer  HO—1 —2 —3 —H Unlinked monomer  Dehydration removes a water molecule, forming a new bond.  HO—1 —2 —3 —4 —H  Longer polymer		(b) Hydrolysis: breaking down a polymer  HO—1—2—3—4—H  Hydrolysis adds a water molecule, breaking a bond.  HO—1—2—3—H  O 2011 Pearson Educaton, Inc.	
Concept Check	Reaction  Glycerol  Fatty ac  H-C-OH	id	H O H O H O H O H O H O H O H O H O H O	H + 0   1   Ra   O
	H O H O O O O O O O O O O O O O O O O O	H O OH R	HO-OH + HO	nosaccharide  Ho-O-O-OH Disaccharide  H20  OH Polysaccharide

	Elements:	
	Functions:	
DDOTEING		nical reactions
PROTEINS	• (lactase) – catalyze chem	ilical reactions
	Defense ()	
	• (milk protein = casein)	
	• (hemoglobin)	
	• Hormones ()	
	•	
	(motor proteins)	
	• (keratin)	
	Monomer =	
	• R group = (vari	able parts of amino acids)
	Properties can vary:	Cide aboin (D. aucum)
	o hydrophobic	Side chain (R group)
	o hydrophilic	R
	o ionic (acids & bases)	α carbon
	• "amino"	н
	a diffilio	N-C-C
	• "acid"	н он
	dciu	Amino Carboxyl
		group group
How to Build	How are <b>peptide bonds</b> formed?	Amino acid Amino acid
Proteins		H B H B
		H-N-¢-C-OH H-N-¢-C-OH H O H O
	Label the peptide bond.	H <sub>2</sub> O
		Polypeptide chain <mark>뷰</mark> 명 뷰 명
		H-N-C-C-N-C-C-OH
		H O H O
Four Levels of	1. Primary Structure	Primary structure
Protein Structure	<b></b>	Amino acids
		acids # G A
	■ 20 different AA's	'H <sub>a</sub> N do the thirds the do the course Lycope
	■link AA's	Amino end
	СНЬ	@ @@@@@@@@@@@@@@@@@
	CH <sub>2</sub> SH	© 15 45 45 45 45 45 45 45 45 45 45 45 45 45
	Peptide bonds are	Primary structure of transthyretin
	formed between	<b><del></del></b>
	Peptide bond	65 55 00 00 00 00 00 00 00 00 00 00 00 00
	during a	
	Side S	115 110 105 100 IOS
	Chains CH <sub>2</sub> SH	120 125
	(condensation)	Carboxyl end
	reaction.	
	(C-terminus)	



NA/I .						
What are						
CHAPERONINS?				_		
How are proteins affected by environment?				Denatured protein		
	change in = change in					
	Denaturation unfolds a protein by disrupting					
	, & & & & &	struct	ture of a protein.			
	Elements:					
NUCLEIC ACIDS	Function:					
	DNA		RNA			
	•stranded helix	•		-stranded		
	• N-bases: A, G, C,		I-bases: A, G, C,			
	• Stores info		Carry info from DN			
	• Longer/larger		RNA, rRNA, mRNA,			
	Sugar:		ugar:			
	Monomer:			NH <sub>2</sub>		
	3 parts:  5' end 5'C - 0 3'C -		Nucleotide	Sugar Dase		
	Nucleoside		Purines	Pyrimidines		
	O 5'C CH <sub>2 O CH<sub>2</sub> O C</sub>		•Adenine •Guanine	•Cytosine •Thymine (DNA) •Uracil (RNA)		
	Phosphate 3'C Sugar		•Double ring	•Single ring		
	(pentose) 3'C (b) Nucleotide 3' end OH (a) Polynucleotide, or nucleic acid			Memory Aid:		
	How are nucleic acids formed?		Formation of phosphodiester bond			
	<ul> <li>Nucleotides join together to form the DN backbone (S-P) via</li> <li>N-bases held together by</li></ul>	IA 	OH Condensation another chemical bond	CH <sub>2</sub> O OH		
				он он		
What is the	Information flow in a cell:		он он			
<b>CENTRAL DOGMA</b> of Biology?	→→					

CARBOHYDRATES	Elements:			
CARBOHTDRATES	Monomers:  Glucose  CH2OH  H CH2OH  C			
	Naming:			
	Role of functional groups:			
	Examples/Types:			
	Functions:  CH <sub>2</sub> OH  H  H  H  H  H  H  H  H  H  H  H  H			
	How are sugars built?  CH2OH  HOH  HOH  HOH  HOH  HOH  HOH  H			
	Role of isomers?  (b) Dehydration reaction in the synthesis of sucrose			
	Elements:			
LIPIDS	Monomers:			
	Types/Examples & Functions:  A(triglyceride):			
	Glycerol + 3 Fatty Acids			
	<ul><li>saturated, unsaturated, polyunsaturated</li><li>and hormones</li></ul>			
	B and hormones C lipid bilayer of cell membrane			
	• tails			



## Protein structure (review) (a) Primary structure (b) Secondary structure (c) Tertiary structure

(d) Quaternary structure

Large Biological Molecules	Components	Examples	Functions	
CONCEPT $5.2$	CH₂OH H ← Q H	<b>Monosaccharides:</b> glucose, fructose	Fuel; carbon sources that can be converted to other molecules or combined into polymers	
Carbohydrates serve as fuel and building material	HOH HOH Monosaccharide monomer	Disaccharides: lactose, sucrose		
and building material		Polysaccharides:  • Cellulose (plants)  • Starch (plants)  • Glycogen (animals)  • Chitin (animals and fungi)	Strengthens plant cell walls Stores glucose for energy Stores glucose for energy Strengthens exoskeletons and fungal cell walls	
CONCEPT 5.3  Lipids are a diverse group of hydrophobic molecules	Glycerol 3 fatty acids	<b>Triacylglycerols</b> (fats or oils): glycerol + 3 fatty acids	Important energy source	
	Head with P 2 fatty acids	Phospholipids: phosphate group + 2 fatty acids	Lipid bilayers of membranes  Hydrophobic tails  Hydrophilic heads	
	Steroid backbone	<b>Steroids:</b> four fused rings with attached chemical groups	Component of cell membranes (cholesterol)     Signaling molecules that travel throug the body (hormones)	
Proteins include a diversity of structures, resulting in a wide range of functions	oteins include a diversity structures, resulting in a		<ul> <li>Catalyze chemical reactions</li> <li>Provide structural support</li> <li>Store amino acids</li> <li>Transport substances</li> <li>Coordinate organismal responses</li> <li>Receive signals from outside cell</li> <li>Function in cell movement</li> <li>Protect against disease</li> </ul>	
Nucleic acids store, transmit, and help express hereditary information	Nitrogenous base Phosphate group P—CH2 O	<ul> <li>Sugar = deoxyribose</li> <li>Nitrogenous bases = C, G, A, T</li> <li>Usually double-stranded</li> </ul>	Stores hereditary information	
anomical of	Sugar Nucleotide monomer	<ul> <li>Sugar = ribose</li> <li>Nitrogenous bases = C, G, A, U</li> <li>Usually single-stranded</li> </ul>	Various functions during gene expression, including carrying instructions from DNA to ribosomes	